Tropical Journal of Pharmaceutical Research January 2017; 16 (1): 165-169 ISSN: 1596-5996 (print); 1596-9827 (electronic) © Pharmacotherapy Group, Faculty of Pharmacy, University of Benin, Benin City, 300001 Nigeria. All rights reserved.

> Available online at http://www.tjpr.org http://dx.doi.org/10.4314/tjpr.v16i1.22

**Original Research Article** 

# Extraction and characterization of radish seed oils using different methods

# Gongling Zhao<sup>1</sup>\*, Yupeng Ren<sup>2</sup> and Hanjun Ma<sup>1</sup>

<sup>1</sup>School of Food Science, Henan Institute of Science and Technology, Xinxiang 453003, <sup>2</sup>Henan Vocational College of Chemical Technology, Zhengzhou 450042, China

\*For correspondence: **Email:** hnzgl@163.com

Received: 24 September 2016

Revised accepted: 18 December 2016

# Abstract

**Purpose:** To evaluate the impact of three different extraction methods on oil yield, physicochemical properties and bioactive ingredients of radish seeds.

**Methods:** Radish seed oil was prepared by traditional solvent extraction (SE), supercritical carbon dioxide extraction (SCE) and sub-critical propane extraction (SPE). The yield, physicochemical properties, fatty acid composition and oxidative stability of the oil extracts were compared. The contents of tocopherol and sulforaphene in the oils were also determined.

**Results:** The oil yield obtained by SPE, SE, SCE were 33.69, 27.17 and 24.10 %, respectively. There were no significant differences in physicochemical properties and fatty acid composition of oils extracted by the three methods. However, SCE oil had the best oxidative stability, and highest contents of vitamin E and sulforaphene, followed by oils from SPE and SE

**Conclusion:** SCE is highly selective for tocopherol and sulforaphene, which could explain its high oil oxidative stability. These results suggest that of the three extraction methods, SCE is best suited for preparing medicinal radish seed oil.

*Keywords:* Radish seed oil, Different extraction methods, Fatty acid composition, Tocopherol, Sulforaphene

Tropical Journal of Pharmaceutical Research is indexed by Science Citation Index (SciSearch), Scopus, International Pharmaceutical Abstract, Chemical Abstracts, Embase, Index Copernicus, EBSCO, African Index Medicus, JournalSeek, Journal Citation Reports/Science Edition, Directory of Open Access Journals (DOAJ), African Journal Online, Bioline International, Open-J-Gate and Pharmacy Abstracts

# INTRODUCTION

There is sufficient scientific evidence indicating that consumption of radish (Raphanus sativus L.) has positive influence on reduction of risks of a number of cancers and cardiovascular diseases, its content of some beneficial due to phytochemicals [1-3]. In traditional Chinese medicine, radish seed oil, which is rich in sulforaphene is used to improve intestinal function and for treatment of digestive problems [4]. Studies have shown that oil yields from radish seeds can be as high as 35-40 %, making them potential new oil sources for food, medicine and cosmetic industries [5].

The quality and safety of natural extracts used as functional foods and medicines must conform with strict regulations related to residual levels of solvents. This necessitates the development and application of new extraction methods in food and pharmaceutical industries [6]. Super-critical carbon dioxide extraction (SCE) and sub-critical propane extraction (SPE) have been widely accepted and used in the extraction of oils and phytochemicals [7,8]. However, there are no reports so far on the use of these techniques for preparation of radish seed oil. In this study, SE, SCE and SPE were used to extract oil from radish seeds. Oil vields, as well as

physicochemical properties and bioactive ingredients of the oils were compared.

## **EXPERIMENTAL**

## Materials and chemicals

Radish seeds were purchased from Wuhan Vegetable Research Institute (Wuhan, China). Standard fatty acid methyl esters, tocopherols ( $\alpha$ ,  $\beta$ -,  $\gamma$ -, and  $\delta$ -isomers) and sulforaphene were procured from Sigma Chemical Co. (St Louis, MO). All other reagents were of analytical grade.

## Determination of proximate composition

The content of moisture, ash, crude lipid, crude protein and crude fiber in the radish seeds were determined by the AOCS Official Methods [9] and expressed on a wet weight basis.

## Solvent extraction (SE)

Radish seed powder was prepared by grinding portions of the seeds in a clean coffee grinder for 15 s. The radish seed powder (1000 g) was subsequently soaked in 5000 mL of n-hexane for 5 h at 25 °C and filtered. The residue obtained was again extracted with 5000 mL of n-hexane. The supernatants from both steps were collected, combined and concentrated by using Rotary evaporator under vacuum at 40 °C to remove n-hexane. The oil obtained was weighed and stored at 4 °C prior to analysis.

## Supercritical carbon dioxide extraction (SCE)

SCE extraction was performed on a 10 L-SFE system (Masson Nem, Guangzhou, China), consisting of an extraction vessel (10 L) and two separators. Radish seed bran powder (1200 g) was put into the extractor and extracted at 26 mPa at 45 °C for 2.5 h. The operating conditions for the two separators were set at 7.5 mPa at 50 °C and 5 mPa at 30 °C, respectively. The flow rate of liquid CO<sub>2</sub> was 35 L/h. The oil in the separators was collected, weighed and stored at 4 °C for analysis.

## Sub-critical propane extraction (SPE)

SPE extraction was performed on a CBE-5L subcritical fluid extraction system (Henan Yalinjie Biological Technology CO., LTD, Anyang, China). The radish seed bran powder (1200 g) was loaded into the extractor and extracted at a pressure of 0.4 mPa and temperature of 40 °C for 2.5 h. Evaporating temperature was set at 40 °C. The resultant oil in the separators was

collected, weighed and stored at 4  $^{\circ}\mathrm{C}$  prior to analysis.

## **Determination of physicochemical properties**

The density, refractive index, saponification value and peroxide value of each of the oils were measured by the IUPAC Methods [10] Oil color was determined according to the AOCS Official Method [11].

## Determination of fatty acid composition

Fatty acid compositions of the oil samples were analyzed according to the IUPAC method 2.302 [12], with an Agilent 6890N gas chromatograph (Agilent Technologies Co., Ltd.), fitted with a flame ionization detector. The column used was a DB-FFAP capillary column; the film thickness was 0.25  $\mu$ m. The length of the column was 30 m and its internal diameter was 0.32 mm. The column, injector, and detector temperatures were set at 180, 230, and 230 °C, respectively. The flow rate of the N<sub>2</sub> carrier gas with a split ratio of 1:20 was set at 70 mL/min.

## Determination of tocopherol composition

Tocopherol ( $\alpha$ -,  $\beta$ - and  $\gamma$ - isomers) contents of the oils were estimated according to the method of Shi *et al* [13], using an LC-10Avp High Performance Liquid Chromatography (Shimadzu, Japan), fitted with a RF-10AXL fluorescence detector (Shimadzu, Kyoto, Japan). Each oil was dissolved in n-hexane and eluted on a silica column (250 × 4.6 mm, 5 µm) (Dalian Yilite, Dalian, China) with n-hexane:isopropyl ether (99 : 1,v/v ratio) at 0.8 mL/min. Column temperature was set at 40 °C.

#### Determination of sulforaphene content

To quantify sulforaphene, a gas chromategraphy-mass spectrometry (GC-MS) was used [14-15]. An Agilent 7890 coupled with an Agilent 5975C mass selective detector and a fused silica capillary column (HP-5, 30 m × 0.25 mm, 0.25 µm of film thickness) were employed. Oven temperature was set at 80 °C for 1 min, then increased to 150 °C at the rate of 20 °C/min and then to 280 °C at the rate of 40 °C/min and held at 280 °C for 3 min. Injector temperature was at 250 °C; carrier gas (helium) was at a flow rate of 1 mL/min; ionization energy was 70 eV; mass range was 40-500 m/z and scan mode was on electron impact. One microliter of the sample was injected with a split ratio of 1:20, and the data were recorded and processed by Xcalibur software.

#### Determination of oil oxidative stability

The oxidative stabilities of the oils were determined on a 743 Rancimat analyzer (Metrohm, Switzerland) [16]. Each oil sample (5 g) was subjected to oxidation at 110  $^{\circ}$ C with air flow rate of 20 L/h. The induction period was recorded automatically.

## RESULTS

#### **Proximate composition**

The proximate composition of radish seed is shown in Table 1 below. The crude lipid content was 35.46 %, which is higher than that of soya bean.

 Table 1: Proximate composition (%) of radish seed

Moisture	Ash	Crude lipid	Crude protein	Crude fiber
6.57	4.71	35.46	18.48	10.05

#### Oil yield

The different extracting methods yielded different levels of oil. Traditional SE method with nhexane for 5 h at 25 °C gave an oil yield of 27.17 % (w/w). The yield from SCE method at 26 mPa in 45 °C for 2.5 h was 24.10 % (w/w) while SPE method at 0.4 mPa in 40°C for 2.5 h was 33.69 % (w/w).

#### **Tocopherol composition**

Results of tocopherol composition of the oils obtained by SE, SCE and SPE showed that  $\gamma$ -tocopherol was the major fatty acid component. The content of  $\gamma$ -tocopherol in SCE oil was as high as 538.90 mg/100 g, while the contents of  $\gamma$ -tocopherol oils from SE and SPE were only 95.76 and 202.86 mg/100 g, respectively. Oil from SE had the highest contents of  $\alpha$ -tocopherol and  $\beta$ -tocopherol, (27.08 and 9.32 mg/100 g respectively), followed by SPE oil (19.32 and 8.38 mg/100 g, respectively). The least was SCE, with 8.85 and 4.26 mg/100 g, respectively.

The variation pattern of  $\delta$ -tocopherol and total tocopherol in the oils was similar to that of  $\gamma$ -tocopherol.

#### Sulforaphene contents

Sulforaphene content of the oil obtained by SCE was the highest (53.34 mg/kg), followed by SPE and SE, which yielded 27.65 and 13.72 mg/kg, respectively.

#### Oil oxidative stability

The induction periods of the oils obtained by SE, SCE and SPE were 4.99, 38.51 and 10.08 h. The oxidative stability of oil from SCE was significantly higher than that from SE or SPE. This may be attributed to the high contents of tocopherol and sulforaphene in the SCE oil.

## DISCUSSION

Traditional extraction methods which use large amounts of toxic solvents to extract plant oils have many drawbacks, such as time factor, residual solvents in oils, and low output rate. Modern extraction methods can solve these problems. Supercritical carbon dioxide extraction (SCE) and subcritical propane extraction (SPE) are the most promising technologies for overcoming the disadvantages of traditional extraction protocols [17]. In this study, three extraction methods (SE, SCE and SPE) were to extract radish seed used oil. The properties physicochemical and bioactive ingredients of the resultant oils were compared. SPE was superior to SCE with respect to extraction yield of oil from radish seeds. This is related to the fact that subcritical propane is a better solvent for triacylglycerols than supercritical carbon dioxide [18,19].

The effect of the extraction methods on traditional physicochemical properties (density, refractive index, saponification value, peroxide value and iodine value) was not significant. However, the peroxide values of SE, SCE and SPE were lower than those generally recommended for commercial vegetable oils [20].

Table 2: Physicochemical properties of radish seed oils obtained by different extraction methods

Parameter	SE	SCE	SPE
Density	0.90	0.92	0.91
Refractive index	1.47	1.47	1.47
Saponification value	182.10	183.90	184.70
Peroxide value (meq/kg)	1.78	0.69	1.68
lodine value	94.64	101.53	99.17
Color (Lovibond, 1 in.)	Y25.00, R2.00	Y30.00, R1.50	Y35.00, R2.60

The color of the oil obtained by SPE was deeper than that of SE or SCE, which implies that it had higher contents of natural pigments (such as carotenoids). The oils obtained by the different extraction methods showed similar fatty acid compositions, with high erucic acid content. which could partly explain the medicinal importance of radish seed oil in Chinese traditional medicine. Although the oil yield of SPE was the highest, its bioactive ingredients (tocopherol and sulforaphene) were lower than those of SCE, which suggests that its extraction potential was lower than that of SCE. The high contents of tocopherol and sulforaphene probably conferred high oxidative stability on the oil obtained by SCE.

## CONCLUSION

The effects of the three extraction methods on traditional physicochemical properties and fatty acid composition of the resultant oils were not significant. Due to its high oxidative stability and selectivity for tocopherol and sulforaphene, SCE appears to be the most suitable method for preparing medicinal radish seed oil.

## DECLARATIONS

#### Acknowledgement

The financial support provided by the Program for Innovative Research Team (in Science and Technology) in University of Henan Province (13IRTSTHN006) is greatly appreciated.

#### **Conflict of Interest**

No conflict of interest associated with this work.

#### **Contribution of Authors**

The authors declare that this work was done by the authors named in this article and all liabilities pertaining to claims relating to the content of this article will be borne by them.

#### **Open Access**

This is an Open Access article that uses a funding model which does not charge readers or their institutions for access and distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by /4.0) and the Budapest Open Access Initiative (http://www.budapestopenaccessinitiative.org/rea d), which permit unrestricted use, distribution, and reproduction in any medium, provided the original work is properly credited.

## REFERENCES

- Kuang P, Song D, Yuan Q, Lv X, Zhao D, Liang H. Preparative separation and purification of sulforaphene from radish seeds by high-speed countercurrent chromatography. Food Chem 2013; 136: 309–315.
- Kuang P, Song D, Yuan Q, Yi R, Lv X, Liang H. Separation and purification of sulforaphene from radish seeds using macroporous resin and preparative highperformance liquid chromatography. Food Chem 2013; 136: 342–347.
- Zhang J, Zhou X, Fu M. Integrated utilization of red radish seeds for the efficient production of seed oil and sulforaphene. Food Chem 2016; 192: 541–547.
- Ahangari B, Sargolzaei J. Extraction of pomegranate seed oil using subcritical propane and supercritical carbon dioxide. Theor Found Chem Eng 2012; 46: 258– 265.
- Wada M, Kido H, Ohyama K, Ichibangase T, Kishikawa N, Ohba Y, Nakashima MN, Kuroda N, Nakashima K. Chemiluminescent screening of quenching effects of natural colorants against reactive oxygen species: Evaluation of grape seed, monascus, gardenia and red radish extracts as multi-functional food additives. Food Chem 2007; 101: 980–986.
- Herrero M, Cifuentes A, Ibañez E. Sub- and supercritical fluid extraction of functional ingredients from different natural sources: Plants, food-by-products, algae and microalgae: A review. Food Chem 2006; 98: 136–148.
- Dos Santos Freitas L, de Oliveira JV, Dariva C, Jacques RA, Caramão EB. Extraction of grape seed oil using compressed carbon dioxide and propane: Extraction yields and characterization of free glycerol compounds. J Agric Food Chem 2008; 56: 2558–2564.
- 8. Firestone D. Official method and recommended practices of the AOCS, 5th edn. AOCS Press, Champaign 1998.
- Paquot C, Hauntfenne A. IUPAC standard methods for the analysis of oils, fats and derivatives. London: Blackwell; 1987.
- Liang S, Yang G, Ma Y. Chemical characteristics and fatty acid profile of foxtail millet bran oil. J Am Oil Chem Soc 2010; 87: 63–67.
- Lim S, Lee J, Kim JK. Analysis of isothiocyanates in newly generated vegetables, Baemuchae (Brassicoraphanus) as affected by growth. Int J Food Sci Tech 2009; 44: 1401–1407.
- Vaughn SF, Berhow MA. Glucosinolate hydrolysis products from various plant sources: pH effects, isolation, and purification. Ind Crop Prod 2005; 21: 193– 202.
- Shi Y, Ma Y, Zhang R, Ma H, Liu B. Preparation and characterization of foxtail millet bran oil using sub-critical propane and supercritical carbon dioxide extraction. J Food Sci and Tech 2015; 52: 3099–3104.
- Oomah, BD, Ladet S, Godfrey, DV, Liang J, Girard B. Characteristics of raspberry (Rubus idaeus L.) seed oil. Food Chem 2000; 69: 187–193.

*Trop J Pharm Res, January 2017; 16(1): 168* 

- 15. Stinco CM, Heredia FJ, Vicario IM, Meléndez-Martínez AJ. Interactions between lipophilic antioxidants measured by photochemiluminescence assay and αtocopherol equivalent antioxidant capacity assay as well as the influence of matrix compounds on the lipophilic antioxidant capacity. LWT-Food Sci Technol 2016; 65: 718–724.
- Reyes N, Reyes I, Tiwari R, Geliebter J. Effect of linoleic acid on proliferation and gene expression in the breast cancer cell line T47D. Cancer Lett 2004; 209: 25–35.
- 17. Harris WS. Linoleic acid and coronary heart disease. Prostag Leukotr Ess 2008; 79: 169–171.
- Lanza M, Ndiaye PM, Tavares FW, Oliveira D, Dariva C, Oliveira JV. Phase Behavior of Castor Oil in Compressed Propane and n-Butane, J Supercrit Fluids2005; 34: 215–221.
- Hamdan S, Daood HG, Toth\_Markus V. Extraction of cardamom oil by supercritical carbon dioxide and subcritical propane. J Supercrit Fluids2008; 44: 25–30.
- 20. Tian G, Li Y, Cheng L, Yuan Q, Tang P, Kuang P, Hu J. The mechanism of sulforaphene degradation to different water contents. Food Chem 2016; 194: 1022–1027.