Original Research Article

Optimized models of xenobiotic-induced oxidative stress in HepG2 cells

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Sent for review: 19 January 2019
Revised accepted: 24 April 2019

Abstract

Purpose: To evaluate the molecular impact of ethanol, sodium selenite, and tert-butyl hydroperoxide (TBHP) on oxidant-antioxidant balance in HepG2 cells to establish an optimized oxidative stress model of HepG2 cells.

Methods: HepG2 cells were treated with ethanol (10 - 500 mM) and sodium selenite (1 - 10 µM) for 24 and 48 h and with TBHP (50 - 200 µM) for 3 and 24 h, respectively. Biomarkers for cellular injury, i.e., lactate dehydrogenase (LDH), aspartate aminotransferase (AST), alanine aminotransferase (ALT), and malondialdehyde (MDA), and for antioxidant system, i.e., superoxide dismutase (SOD), catalase (CAT), and total glutathione content, were determined.

Results: All treatments increased the levels of LDH, AST, ALT, and MDA but decreased SOD and CAT activities and the total glutathione content in HepG2 cells. Oxidative stress was induced by these oxidative stressors in HepG2 cells via oxidant-antioxidant imbalance, with TBHP (100 µM, 3 h) acting as a powerful oxidant based on the minimal time to induce oxidative stress. The antioxidants, ascorbic acid and gallic acid, improved oxidant-antioxidant imbalance against xenobiotic-induced oxidative stress in HepG2 cells.

Conclusion: These oxidative stress models are suitable for investigating the antioxidant and/or hepatoprotective potential of chemicals, including natural compounds.

Keywords: Ethanol, Sodium selenite, Tert-butyl hydroperoxide, Oxidative stressor, Oxidant-antioxidant balance

INTRODUCTION

The liver plays a key role in detoxification, including the regulation of physiological homeostasis hence, damage to this organ results in bodily dysfunction [1]. Toxic compounds are among the most pivotal causes of liver toxicity due to the generation of free radical byproducts, especially reactive oxygen species (ROS) [2]. Imbalance in oxidant-antioxidant systems leads to oxidative stress, which is harmful to cells, tissues, and organs and is associated with diseases, e.g. liver diseases [2].

Xenobiotic exposure is an exogenous factor that account for most ROS generation [3]. Examples
of such xenobiotics include ethanol and sodium selenite, which may be introduced through the diet and lifestyle, and tert-butyl hydroperoxide (TBHP), which is typically related to industrial air pollution [4]. Although these xenobiotics have been used as oxidative stress stimuli, causing hepatotoxicity [5-7] via the induction of cellular oxidative stress [8-9], neither the concentrations nor the trial intervals were optimized in these previous studies.

To understand the mechanism of oxidative stress in the liver, a human hepatoma (HepG2) cell culture has been employed as a liver model due to its reproducible manageability and its similarity in features and functions to human hepatocytes [10]. However, most preliminary studies reported the use of only one or two concentrations of substances to induce oxidative stress in HepG2 cells, without validation of the doses and duration of the treatments. Furthermore, previous reports focused on only one oxidative stressor without comparison with others [7-9]. Therefore, the present study aimed to evaluate the cellular impacts of ethanol, sodium selenite, and TBHP on the oxidant-antioxidant balance in HepG2 cells and to optimize the concentrations and trial intervals of these potential inducers to establish a definitive HepG2 cell oxidative stress model. Moreover, the oxidative stress-inducing capacity of ethanol, sodium selenite, and TBHP was assessed.

EXPERIMENTAL

Chemicals

Sodium selenite, tert-butyl hydroperoxide (TBHP, 70 % w/v), sodium pyruvate, 2,4-dinitrophenylhydrazine (DNPH), 5,5’-dithiobis(2-nitrobenzoic acid) (DTNB), nitroblue tetrazolium (NBT) were purchased from Sigma-Aldrich Chemical (St. Louis, MO, USA). Superoxide dismutase (SOD), catalase (CAT), xanthine oxidase, β-nicotinamide adenine dinucleotide 2’-phosphate (NADPH), glutathione reductase (GR), reduced glutathione (GSH), malondialdehyde (MDA), thiobarbituric acid (TBA), L-aspartate, L-alanine, and α-ketoglutarate were also purchased from Sigma-Aldrich Chemical. Absolute ethanol and gallic acid were purchased from Merck (Darmstadt, Germany). Ascorbic acid was obtained from Rankem (New Delhi, India). Dulbecco’s modified Eagle medium (DMEM, Cat. No. 11885-084), Dulbecco’s modified Eagle medium nutrient mixture F-12 (DMEM/F-12 without phenol red, Cat. No. 21041-025), and fetal bovine serum (FBS) were purchased from Gibco (New York, USA). All other laboratory chemicals were of the highest purity and were obtained from chemical suppliers.

Induction of oxidative stress

HepG2 cells (HB-8065) were purchased from ATCC (Manassas, Virginia, USA) and cultured in DMEM supplemented with 10 % FBS, Glutamax®, penicillin, and streptomycin under 95 % humidity in an atmosphere of 5 % CO2 at 37 °C. The cells were cultured in DMEM in a 6-well plate (5 x 10⁵ cells/well). The medium was discarded after 48 h, and the cells were washed with 1×PBS and replaced with DMEM/F-12 without phenol red supplemented with 10 % FBS, Glutamax®, penicillin, and streptomycin containing ethanol, sodium selenite, or TBHP at various concentrations and incubated for various intervals: ethanol was used at 10, 100, and 500 mM and sodium selenite at 1, 5, and 10 μM for 24 or 48 h; TBHP was used at 50, 100, and 200 μM for 3 or 24 h if not otherwise indicated. The control was incubated with DMEM/F-12 [8,9].

Determination of lactate dehydrogenase (LDH) activity

The toxicity of ethanol, sodium selenite, and TBHP was assessed in culture medium using an LDH Assay Kit (Sigma®). The percentage of toxicity was calculated based on LDH activity by comparison with 2 % (v/v) Triton X-100.

Determination of aspartate aminotransferase (AST) and alanine aminotransferase (ALT) levels

The culture medium was incubated with AST (L-aspartate and α-ketoglutarate) or ALT (L-alanine and α-ketoglutarate) substrates at 37 °C for 30 or 20 min, respectively, and DNPH was added for 20 min before the addition of 4 N NaOH. Absorbance was measured at a wavelength of 505 nm. Aspartate aminotransferase (AST) and alanine aminotransferase (ALT) levels were expressed as international units per liter (IU/L) by comparison with a standard curve of sodium pyruvate [11].

Determination of lipid peroxidation by the thiobarbituric acid reactive substances (TBARS) assay

HepG2 cell homogenate was mixed with 10 % trichloroacetic acid (TCA) followed by centrifugation at 2,300×g at 4 °C for 10 min. The supernatant was pipetted to a new tube, and 0.8 % TBA was added before boiling at 100 °C for 15 min, followed by immediate cooling. The MDA in the supernatant was measured by
spectrofluorometry at excitation and emission wavelengths of 520 and 590 nm, respectively, and by comparison with the standard curve of MDA [12].

**Determination of superoxide dismutase (SOD) activity**

HepG2 cell homogenate was mixed with chloroform and ethanol (1:1.67) before centrifugation at 14,000×g at 4 °C for 30 min. The supernatant was incubated with a reaction mixture containing 0.6 mM ethylenediaminetetraacetic acid (EDTA), 3 mM xanthine, 3 mM NBT, 400 mM Na2CO3, and 1 g/L bovine serum albumin (BSA) at 25 °C for 20 min, followed by the addition of xanthine oxidase. After incubation, 0.8 mM CuCl2 was added. Absorbance was measured at a wavelength of 550 nm and converted to SOD activity by comparison with an SOD standard [12].

**Determination of catalase (CAT) activity**

HepG2 cell homogenate was incubated with 130 μM H2O2 at 37 °C for 1 min. The reaction was terminated by the addition of 32.4 mM ammonium molybdate, followed by absorbance measurement at a wavelength of 405 nm. The CAT activity was determined by comparison with a CAT standard [12].

**Assessment of the total glutathione (GSH) content**

HepG2 cell homogenate was mixed with a mixture of 0.01 M potassium phosphate buffer (pH 7.0), GR, and 1.5 mg/mL DTNB, followed by the addition of 0.16 mg/mL NADPH. The absorbance at a wavelength of 405 nm was recorded every 30 s for 10 min. The total GSH content was calculated by comparison with the slope of the GSH standard curve [12].

**Statistical analysis**

The results are expressed as the mean ± SD. All data were statistically analyzed using one-way analysis of variance (ANOVA) followed by Tukey’s post hoc test. *p < 0.05* was considered statistically significant.

**RESULTS**

**Effects of ethanol, sodium selenite, and TBHP on HepG2 cell cytotoxicity**

The LDH activity over 10 is considered cytotoxic [13]. No cytotoxicity was observed in the treatment of HepG2 cells with ethanol (10, 100, and 500 mM) for 24 h, although treatment for 48 h at the two higher concentrations resulted in significant cytotoxicity (Figure 1 A). In contrast, significant cytotoxicity was observed after treating HepG2 cells with sodium selenite (1, 5, and 10 μM) for 24 and 48 h (Figure 1 B) and with TBHP (50, 100, and 200 μM) for 3 and 24 h (Figure 1 C). These observations suggest that compared with sodium selenite and ethanol, TBHP has the highest toxic potential due to the minimal time required to induce cytotoxicity.

**Effects of ethanol, sodium selenite, and TBHP on cellular antioxidant system**

Oxidative stress is indicated by decreases in SOD and CAT activities, including the depletion of total GSH [5]. Exposing HepG2 cells to

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ethanol (Figure 2A, 2D, and 2G) and sodium selenite (Figure 2B, 2E, and 2H) for 24 and 48 h and to TBHP for 3 and 24 h (Figure 2 C, F, and I) resulted in significant dose- and time-dependent decreases in SOD and CAT activities as well as in the total GSH content.

MDA is the end product of lipid peroxidation, and its level indicates cell damage [20]. Treating HepG2 cells with ethanol (Figure 2J) and sodium selenite (Figure 2 K) for 24 and 48 h and TBHP (Figure 2 L) for 3 and 24 h significantly elevated MDA levels in a dose- and time-dependent manner. These observations suggest that TBHP is the strongest oxidative stressor due to its ability to disrupt the oxidant-antioxidant balance, decrease SOD and CAT activities, and deplete total GSH with an increase in MDA within 3 h.

**Effects of ascorbic acid and gallic acid on cytotoxicity and cellular antioxidant system**

Ascorbic acid at 25 - 100 µM and gallic acid at 29 - 294 µM were not toxic to HepG2 cells. The oxidative stressors ethanol, sodium selenite, and TBHP caused cytotoxicity in HepG2 cells, and cotreatment with either ascorbic acid or gallic acid indicated a tendency to ameliorate LDH toxicity to the cells (Figure 3 A).

Ethanol, sodium selenite, and TBHP significantly reduced cellular SOD (Figure 4 A) and CAT (Figure 4 B) activities followed by depletion of total GSH (Figure 4 C). Ascorbic acid and gallic acid markedly increased cellular antioxidant parameters, namely, SOD and CAT activities and the total GSH content.

Neither ascorbic acid nor gallic acid changed the MDA level in HepG2 cells but the MDA levels were significantly increased after treatment with ethanol, sodium selenite, and TBHP for 24 h (Figure 4 D). When HepG2 cells were treated with ethanol, sodium selenite, and TBHP in combination with ascorbic acid and gallic acid for 24 h, a significant decrease in the MDA level was observed. These findings support the approach of inducing oxidative stress in HepG2 cells using ethanol, sodium selenite, and TBHP and reveal the protective antioxidant potential of ascorbic acid and gallic acid against oxidative stress in these HepG2 cell models.

**DISCUSSION**

Excessive production of ROS leads to cellular injury, causing cell membrane rupture and loss of selective membrane permeability, which result in...
Figure 4: Effect of ascorbic acid and gallic acid on the cellular antioxidant system in ethanol-, sodium selenite-, and TBHP-induced oxidative stress in HepG2 cell models. A) SOD activity, B) CAT activity, C) total GSH content, and D) MDA level. *p < 0.05 vs the controls; **p < 0.05 vs treatment with ethanol, sodium selenite, or TBHP alone

Antioxidants are composed of two components, one of which involves major biochemical defense enzymes, such as SOD and CAT. The first mechanism of defense against free radical is SOD which transforms superoxide anion (O$_2^-$) to hydrogen peroxide (H$_2$O$_2$), followed by CAT, which catalyzes H$_2$O$_2$ to water and oxygen [5]. The other component is a non-enzymatic GSH system that is mostly synthesized in the liver. The predominant mechanism of GSH against free radicals is that it acts as a hydrogen or electron donor in detoxification [16]. In the present study, all oxidative stressors injured HepG2 cells via several pathways leading to the release of LDH, AST, and ALT, elevation in the TBARS levels, decreases in SOD and CAT activities and the depletion of total GSH content.

Alcohol dehydrogenase (ADH) is the main enzyme that metabolizes ethanol to acetaldehyde, which is subsequently converted by aldehyde dehydrogenase to acetate, a molecule that is toxic to organs and to systemic circulation [17]. The microsomal ethanol oxidation system (MEOS) and CAT contribute to the metabolism of ethanol, including the production of ROS [18]. These results were consistent with those of a previous study showing that ethanol at a dose of 50 mM for 24 h resulted in decreases in SOD and CAT activities and GSH content and an increase in MDA levels in HepG2 cells [19]. Yao et al. (2007) reported that ethanol at a dose of 100 mM for 24 h increased the AST, LDH, and MDA levels with a decrease in the GSH level in human hepatocytes [9]. In this study, ethanol concentrations up to 500 mM were used to treat HepG2 cells to demonstrate a change in the cellular antioxidant system.

Sodium selenite is a dietary form of selenium, but it is classified as a toxic compound. The hepatic glutathione system converts sodium selenite to selenium and consequently produces ROS, mainly O$_2^·$ and H$_2$O$_2$. Although selenium is an essential element, previous studies have suggested that selenium from the metabolism of sodium selenite influences oxidative stress-related cytotoxicity in HepG2 cells [8]. In accordance with this study, a previous study demonstrated a decrease in GSH content and LDH leakage in HepG2 cells treated with 10 μM sodium selenite for 24 h [7]. Furthermore, treatment with sodium selenite at concentrations ranging from 5 to 25 μM for 24 h resulted in an increase in LDH leakage and ROS production and a decrease in GSH content in HepG2 cells [8].

A short-chain organic hydroperoxide used as a bleaching and oxidizing agent in polymerization in industrial processes is TBHP [20]. In the liver, TBHP is metabolized via cytochrome P450 2E1, which consequently generates ROS, mainly peroxy and alkoxyl, and leads to the initiation of oxidative stress [6]. Alternatively, TBHP is converted by the glutathione system to tert-butyl alcohol and glutathione disulfide, resulting in glutathione depletion [5]. In previous studies, the treatment of HepG2 cells with a dose of TBHP greater than 50 μM for 3 h elevated the TBARS level and decreased SOD and CAT activities [21]. In addition, HepG2 cells treated with 200 μM TBHP for 3 h showed a decrease in GSH content with a reduction in SOD and CAT activities, followed by an increase in MDA level and LDH leakage [21]. Oxidative stress was observed in HepG2 cells after treatment with 250 μM TBHP for 3 h. The viability of TBHP-treated HepG2 cells was 47.33 %, indicating cell growth suppression, which is related to increases in ROS and MDA production along with GSH depletion [22].

The present observations suggest that the cellular oxidant-antioxidant system in HepG2 cells was impaired by the excessive production of ROS due to the metabolism of ethanol, sodium selenite, and TBHP, resulting in oxidative stress. However, the potency of each substances to
induce oxidative stress was not equivalent, and the concentration and time of incubation are important factors. As it requires the shortest exposure time, TBHP might be the strongest oxidative stressor. Further study of these three compounds was carried out to assure an efficient oxidative stress model. Ethanol, sodium selenite, and TBHP were evaluated in combination with ascorbic acid and gallic acid, both of which are known as strong natural antioxidants.

Structures of gallic acid and ascorbic acid contain several hydroxyl groups, which are important in direct reactions with free radicals and exhibit a potent radical-scavenging effect [23] via a hydrogen- or an electron-donating mechanism to eliminate an unpaired electron and neutralize free radicals [24]. In this study, oxidative stress was observed in HepG2 cells treated with ethanol, sodium selenite, and TBHP. Suppression of SOD and CAT activities and depletion of total GSH in HepG2 cells occurred. Ascorbic acid and gallic acid sustain the intracellular antioxidant system via electron-donating mechanisms and terminate the oxidative chain reaction, resulting in the inhibition of either initial or continued oxidative stress [25]. Consequently, both compounds were able to restore SOD and CAT activities and reverse the depletion of the total GSH content. These observations support these oxidative stress models as efficient systems for further study of the antioxidant activity of chemical/natural antioxidants.

**CONCLUSION**

In this study, ethanol-, sodium selenite-, and TBHP-induced oxidative stress models in HepG2 cells have been established. Though the required concentration of TBHP (100 µM) is higher than that of sodium selenite (10 µM), it exerts the strongest oxidative stimulus, requiring the minimal time (3 h) to induce oxidative stress. Even ethanol acts as a mild oxidative stressor and might be the most relevant compound for daily exposure. Hence, to better understand the occurrence of chronic disease-associated oxidative stress, ethanol (100 mM, 24 h) is an optimal oxidative stressor. Nonetheless, as these oxidative stressors differ in their generation of ROS and involvement in oxidative pathways, a given oxidative stressor should be carefully considered when choosing the optimal stressor to ultimately draw a precise evaluation.

**DECLARATIONS**

**Acknowledgements**

This work was supported by Research Group for Pharmaceutical Activities of Natural Products using Pharmaceutical Biotechnology, Faculty of Pharmaceutical Sciences, Khon Kaen University, Thailand under Grant PANPB2561. Yollada Sriset acknowledges the Graduate School, Khon Kaen University for scholarship (no. 601H215).

**Conflict of interest**

No conflict of interest is associated with this study.

**Contribution of authors**

The authors declare that this work was done by the authors named in this article and that all liabilities pertaining to claims relating to the content of this article will be borne by them. Yollada Sriset performed the experimental work and statistical analysis, prepared the figures, and drafted the manuscript. Waranya Chatuphonprasert planned the experimental work, verified the data, supervised the whole study, and revised the manuscript. Kanokwan Jarukamjorn designed the study, participated in the conceptual discussion, and revised the manuscript. The manuscript was thoroughly read and approved by all the authors for publication.

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